NATIONAL TRANSPORTATION SAFETY BOARD

Office of Research and Engineering Materials Laboratory Division Washington, D.C. 20594

9/21/2015



Report No. 15-101

MATERIALS LABORATORY FACTUAL REPORT

A. ACCIDENT INFORMATION

Place: Washington, DC

Date : January 12, 2015 3:15 p.m. EST Vehicle : WMATA train #302 Yellow Line

NTSB No. : DCA15FR004 Investigator : Robert Gordon

B. COMPONENTS EXAMINED

1) Sample of water taken from the leak through the tunnel liner.

2) Sample of water collected from the tunnel floor opposite the arcing location.

3) Sample of cable insulation from the incident cables.

C. DETAILS OF THE EXAMINATION

The water samples collected were tested for pH in the NTSB laboratory. Sample #1 measured 7.4 on the pH scale and sample #2 measured 7.6 on the pH scale. Further tests on the water samples, including ion chromatography¹, conductivity and resistivity² were performed by IMR and the results are attached to this report.

The sample of insulation was collected from one of the cables removed from the incident site. This sample was sent to IMR for thermogravimetric analysis (TGA) of the component layers. The results of that analysis are attached to this report.

Joseph Panagiotou Fire & Explosion Investigator

¹ Method: CAP-017P (ICP-AES)

² Method: ASTM D 1125-14 Method A

131 Woodsedge Drive Lansing, NY 14882 T: 1.607.533.7000 | F: 1.607.533.9210

TEST REPORT

IMR Report Number 201506245A

May 27, 2015

www.imrtest.com

Nancy McAtee National Transportation Safety Board 490 L'Enfant Plaza Washington, DC 20594

SUMMARY

Two samples were received for Ion Chromatography analysis

The results are below.

PO Number Credit Card

Date Received May 14, 2015

Sample#

1-Water sample (flow thru tunnel)

2 – Water sample (opposite arcing location)

Sample ID DCA15FR004

CHEMISTRY

Analyte	Sample #1	0.2	
Bromide	0.1		
Chloride	273	213	
Fluoride	0.3	0.3	
Nitrate	<0.1	<0.1 2.0	
Nitrite	<0.1 <0.1		
Phosphate	<0.1 <0.1		
Sulfate	14.2	10.5	

Results in parts per million unless otherwise indicated.

Method(s): Modified ASTM D 4327-11 (IC)

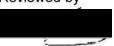




Reviewed by



Lisa Wackowicz Technical Operations Coordinator Reviewed by



Brian Wackowicz Senior Chemist

All procedures were performed in accordance with the IMR Quality Manual, current revision, and related procedures: and the PWA MCL Manual F 23 and related procedures. The information contained in this test report represents only the material tested and may not be reproduced, except in full, without the written approval of IMR Test Labs ("IMR"). IMR maintains a quality system in compliance with the ISO/IEC 17025 and is accredited by the American Association for Laboratory Accreditation (A2LA), certificates #1140.01 and #1140.02. IMR will perform all testing in good faith using the proper procedures, trained personnel, and equipment to accomplish the testing required. IMR's liability to the customer or any third party is limited at all times to the amount charged for the services provided. All samples will be retained for a minimum of 6 months and may be destroyed thereafter unless otherwise specified by the customer. The recording of false, fictitious, or fraudulent statements or entries on this document may be punished as a felony under federal statutes. IMR Test Labs is a GEAE S-400 approved lab (Supplier Code T3983).

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Original Date September 10, 2015

Revision Date September 11, 2015

Nancy McAtee National Transportation Safety Board 490 L'Enfant Plaza Washington, DC 20594

PO Number Credit Card

Date Received August 28, 2015

Sample# 1, 2

TEST REPORT

IMR Report Number 201510929 - Revision 1 (Additional Testing)

SUMMARY

Two samples were received for chemical analysis, conductivity and resistivity testing.

A 5 mL aliquot was taken of each sample and placed in individual beakers. The samples were taken to dryness on a hotplate. The resulting residue was digested with mixed acids and then brought up to a volume of 100 mL. The solutions were then analyzed by ICP-AES.

Conductivity and resistivity were determined on the "as received" sample using ASTM D 1125-14. The conductivity measurement was made at 25.0° C. The resistivity was calculated from the sample's conductivity reading.

The results are on the following page(s).





Reviewed by



Andy Ensign for Lisa Wackowicz Technical Operations Coordinator Reviewed by



Brian Wackowicz Senior Chemist

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CHEMISTRY

Element	1	2	
Ca	0.0078	0.0088	
Fe	0.0021	0.0120	
Mg	0.0017	0.0021	
Na	0.0130	0.0110	
Si	0.0018	0.0053	

Other elements tested (<0.0005%): Ag, Al, As, Au, B, Ba, Be, Bi, Cd, Ce, Co, Cr, Cu, In, K, Li, Mn, Mo, Nb, Ni, P, Pb, Pd, Pt, Sb, Se, Sn, Sr, Ta, Te, Ti, Tl, V, W, Zn, & Zr.

Results in weight percent unless otherwise indicated.

Method(s): CAP-017P (ICP-AES)

CONDUCTIVITY AND RESISTIVITY

Parameter	1	2	
Conductivity	1,154 µmho/cm	1,160 µmho/cm	
Resistivity	866.6 Ω·cm	862.1 Ω·cm	

Conductivity result in μ mho/cm (micromhos/centimeter) and resistivity result in Ω -cm (ohm centimeters). Note that μ mho/cm and μ S/cm (microsiemens/centimeter) are equivalent.

Method(s): ASTM D 1125-14 Method A

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May 27, 2015

Nancy McAtee National Transportation Safety Board 490 L'Enfant Plaza Washington, DC 20594

TEST REPORT

IMR Report Number 201506245B

PO Number

Credit Card

Date Received

May 14, 2015

Sample#

assembly piece

Sample ID DCA15FR004

SUMMARY

One cable insulation assembly was received for TGA analysis of the component layers. The insulation assembly (Figure 1) consists of an inner upjacket (which would have cover the conductors), a middle insulation layer and an outer jacket layer.

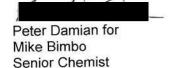
3 - Electrical cable insulation TGA results are reported for each layer.

The results are on the following page(s).





Reviewed by



Reviewed by



Rob Demaree, Manager Chemistry Department

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THERMOGRAVIMETRIC ANALYSIS (TGA)

TGA analysis was performed per ASTM E 1131-08. A sample was heated at a rate of 10°C/min from 30°C to 850°C in in air. Each sample decomposed according to a profile. The initial decomposition was characterized by noting the temperatures at which the samples lost 1%, 2%, 5% and 10% of their initial mass. At the conclusion of the run the sample's noncombustible filler content is also noted.

Individual results are presented in Figures 2-4.

Sample	1% mass loss Temperature	2% mass loss Temperature	5% mass loss Temperature	10% mass loss Temperature	Noncombustible Filler (%)
Inner Upjacket	261°C	313°C	342°C	359°C	47.5
Middle Insulation	270°C	310°C	381°C	414°C	37.0
Outer Jacket	182°C	262°C	314°C	339°C	5.4



Figure 1: Image of the cable insulation assembly, showing the three layer materials.

Figure

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TGA results for the inner upjacket material of the cable insulation assembly

Sample: DCA15 FR004 Upjacket

Size: 7.3320 mg

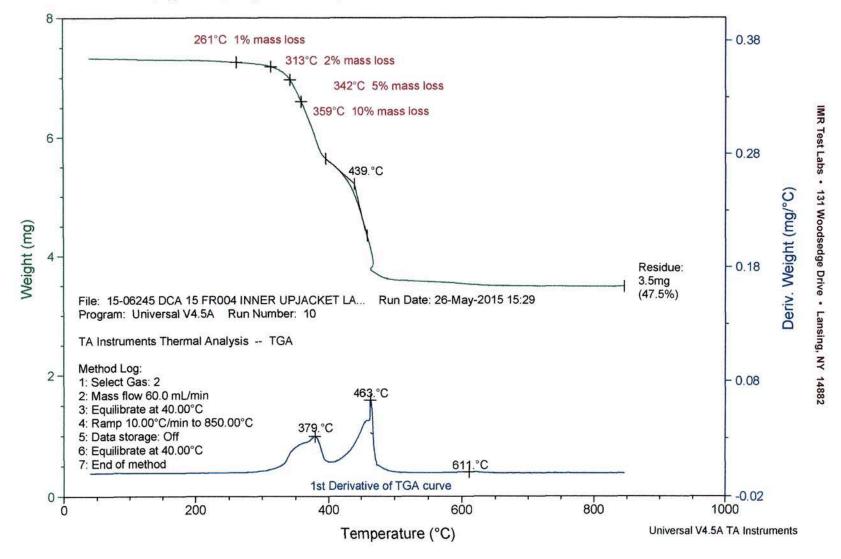
Method: ASTM E1131 in air to 850C

Comment: DCA 15 FR004, upjacket inner layer, air to 850, pan 3

File: 15-06245 DCA 15 FR004 INNER UPJACKET ... **TGA**

Operator: RJD CSDR 7485 Run Date: 26-May-2015 15:29

Instrument: TGA Q500 V6.7 Build 203



Figure

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TGA results for the middle insulation material of the cable insulation assembly.

Sample: DCA15 FR004 Insulation

Size: 15.5890 mg

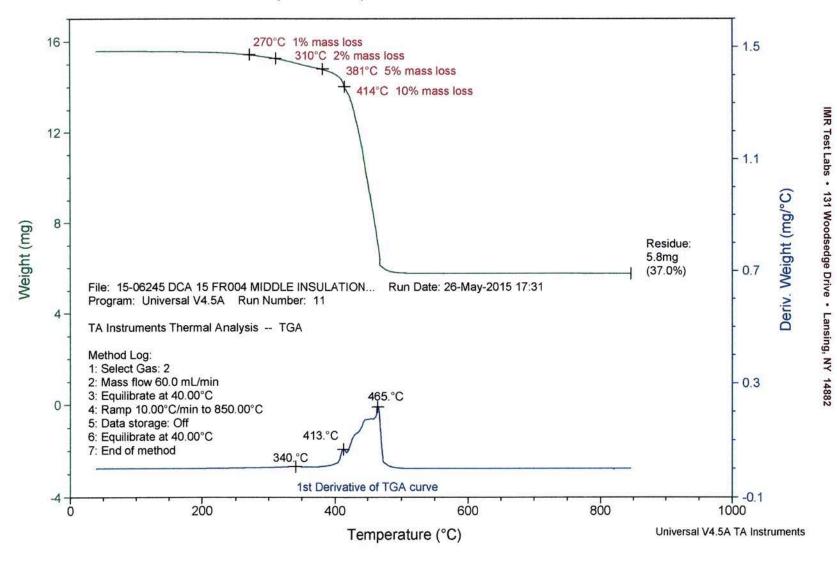
Method: ASTM E1131 in air to 850C

Comment: DCA 15 FR004, middle Insulation layer, air to 850, pan 4

File: 15-06245 DCA 15 FR004 MIDDLE INSULATI... **TGA**

Operator: RJD CSDR 7485 Run Date: 26-May-2015 17:31

Instrument: TGA Q500 V6.7 Build 203



Figure

TGA results for the outer jacket material of the cable insulation assembly.

Sample: DCA15 FR004 Outer Jacket

Size: 4.5390 mg

Method: ASTM E1131 in air to 850C

Comment: DCA 15 FR004, outer jacket layer, air to 850, pan 5

File: 15-06245 DCA 15 FR004 OUTER JACKET.001 **TGA**

Operator: RJD CSDR 7485 Run Date: 26-May-2015 19:31

Instrument: TGA Q500 V6.7 Build 203

